AC/27/06/2023/RS1



RISE WITH EDUCATION

NAAC REACCREDITED - 'A' GRADE

SIES College of Arts, Science and Commerce (Autonomous)

Affiliated to University of Mumbai

Syllabus under NEP effective from June 2023

Programme: B.Sc.

Subject: Biochemistry

Vocational Skill Enhancement Course (VSC)

Class: F.Y.B.Sc. Semester: I and II

Choice Based Credit System (CBCS)

Content

1.	Preamble	3
2.	Overall Credit Structure for F.Y.BSc	.4
3.	Credit Structure of courses offered by Biochemistry Dept. (Semester I & II)	5
1	Cyllabora for VCC Degis to alg 0 to sharing up in Disch aminture	_
4.	Syllabus for VSC- Basic tools & techniques in Biochemistry	0
5.	References for VSC	7
٠.		
6.	Scheme of examination	7

Preamble

Biochemistry is a branch of biological science that delves into the chemical processes and information pathways governing the survival and propagation of life. It is an interdisciplinary science providing the learner an opportunity to elucidate molecular mechanisms and explore the intricate world of biomolecules and their applications. Under the aegis of New Education Policy-2020, the department offers a three-year BSc program and a four-year BSc Honours program with Biochemistry major or minor. At the first year level, along with core Biochemistry course, the department also offers Vocational skill enhancement course and Skill enhancement course. In this program, we will embark on a comprehensive journey from the structure and function of biomolecules to their interactions and implications in health and disease. Through lectures, laboratory sessions, and interactive discussions, the student will not only gain insights of the biochemical processes and pathways, but also develop skills for employability and an aptitude for research.

Objectives:

The goal of this interdisciplinary Biochemistry program is

- 1. Foundational knowledge: To build the foundation of Biochemistry and encourage the student to pursue Biochemistry at higher level.
- 2. Application of Biochemistry: To enable the student to recognize the application of biochemistry in areas of nutrition and food, pharmaceuticals, diagnostics, clinical research, bioinformatics, forensics, etc.
- 3. Laboratory skills: To develop essential laboratory skills for the experimental analysis of biochemical principles.

Program Outcome

At the end of the first year, the student should be able to:

- Comprehend the concepts in nutrition and importance of proper nutrition thus laying a foundation for the field of nutrition and dietetics
- Understand the physical and chemical properties of biomolecules
- Co-relate the structure of biomolecules with their properties and functions
- Explain the role of biomolecules in maintaining structural integrity as well as their role in cellular pathology
- Apply the experimental skills in studying biomolecules and cellular processes
- Recognize the application of biochemistry in diverse fields of food, nutrition, clinical research, drug discovery, diagnostics, forensics, genomics, proteomics and bioinformatics.
- Use basic computational skills in documentation and scientific data presentation.

Evaluation: Student's understanding of biochemistry will be evaluated through a combination of examinations, quizzes, laboratory reports, and class participation. These assessments are designed to gauge learner's comprehension of both theoretical concepts and practical applications.

Overall Credit Structure for F.Y. B.Sc.

Semester	Core I	Core	OE	VSC, SEC (VSEC)	AEC, VEC, IKS	OJT, FP, CEP, CC	Credits/ semester	Degree/ cumulative credits
I	4C	4C	4C	VSC- 2C SEC- 2C	AEC- 2C VEC- 2C IKS- 2C	Nil	22C	UG
II	4C	4C	4C	VSC- 2C SEC- 2C	AEC- 2C VEC- 2C	CC- 2C	22C	Certificate 44C
Total Credits	8C	8C	8C	8C	10C	2C	44C	

OE: Open Elective/Generic open elective

VSC: Vocational Skill Course SEC: Skill Enhancement Course AEC: Ability Enhancement Course VEC: Value Education Course IKS: Indian Knowledge System

OJT: On-job training FP: Field Project

CEP: Community engagement and service

CC: Co-curricular courses

Credit Structure of courses offered by Biochemistry department for F.Y. B.Sc. Biochemistry

Name of Program: Biochemistry		B.Sc. Biocher	mistry N	Name of Department:			
Class	Semester	Course Code	Course Title	Credits	No. of lectures/ per week	Marks	
		SIUBCCC111	Foundations of Biochemistry	03	03	75	
		SIUBCCCP111	Biochemistry Practical	01	02	25	
FYBSc	I	SIUBCVS111	Basic tools &	02	03		
			techniques in	(01Th +	(01Th +		
			Biochemistry	01P)	02P)	50	
		SIUBCSE111	Good Laboratory	02	03		
			Practices & Soft	(01Th +	(01Th +		
			Skills	01P)	02P)	50	
		SIUBCCC121		03	03	75	
			Cell Biology				
		SIUBCCCP121	Cell Biology Practical	01	02	25	
FYBSc	II	SIUBCVS121	Basic tools &	02	03		
			techniques in	(01Th +	(01Th +		
			Biochemistry	01P)	02P)	50	
		SIUBCSE121	Good Laboratory	02	03		
			Practices & Soft	(01Th +	(01Th +		
			Skills	01P)	02P)	50	

SI: SIES

U: Undergraduate BC: Biochemistry CC: Core Course

VS: Vocational Skill Course SE: Skill Enhancement Course

Semester I and II Syllabus- Vocational Skill Enhancement Course

Course Title: Basic tools & techniques in Biochemistry

Course code: SIUBCVS111 and SIUBCVS121

Credits: 02 Hours/week: 01L + 02 P

Course Outcome:

On completing the Course, the learner should be able to

- 1. Understand the various units of concentration. Calculate and interconvert the units of concentration.
- 2. Skilfully carry out serial dilution of a sample.
- 3. Prepare buffers of particular pH and verify using pH paper and pH meter
- 4. Identify the components and understand the working of pH meter
- 5. Comprehend the principle and employ chromatography for separation of biomolecules
- 6. Understand the principle and working of a centrifuge. Apply the principle in separation by density gradient.
- 7. Understand the interaction of electromagnetic radiations with mater. Identify the components of a colorimeter.
- 8. Verify if a coloured solution obeys Beer Lambert law
- 9. Determine the concentration of analyte using Beer-Lambert law by calculation and graphically.

Sr no	Title	Theory (T)/ Practical (P)
1.	Methods of expressing the concentration of a given solution (mole, molarity, normality, percent solution, ppm & ppb)	T
2.	Introduction to the concept of serial dilution.	T + P
3.	Preparation of buffers	P
4.	a. Preparation of a red cabbage/ turmeric pH indicator.b. Study the components of a pH meter.c. Determine the pH of a solution using a pH strip & a pH meter.	T + P
5.	Chromatographic separation of sugars present in fruit juice (ascending paper chromatography)	T + P
6.	a. Principle, components, and working of a centrifuge- Bench top and high-speed cold.b. Use of density gradient centrifugation.	T + P
7.	a. Introduction to basics of spectroscopy.b. Components of a colorimeterc. Verification of Beer-Lambert Law	T + P
8.	Extraction of pigments from flowers and determination of lambda max	Р
9.	Phlebotomy (demonstration)	T + P

REFERENCES FOR VSC

- 1. Biochemical Methods Edition 3 by Sadasivam and Manickam.
- 2. An Introduction to Practical Biochemistry, 3rd edition- David T. Plummer
- 3. Microbiology, 5th edition- Michael Plczar Jr, E.C.S Chan, Noel Krieg.
- 4. Biophysical Techniques- Upadahyay, Upadhyay & Nath
- 5. Research Methodology- C.R Kothari
- 6. Butterfield Jeff Soft Skills for everyone.
- 7. Communication Skills for Engineers & Scientists- Sharma, Sangita and Binod Mishra.

Scheme of Examination:

Credits	Course Type	Distribution of Credits	Sem end	Internal	Practical	Total
2	VSC	without sem end exam		50		50

Details for Internal Assessment:

Weightage for Internal (marks)	Min. marks required for passing	Pattern of Evaluation
VSC: 50 marks (without sem end exam)	20	20 marks- class test (No retest) + 30 marks- Open to Department OR 50 marks from Practical- journal+ viva+ exam etc (continuous evaluation)

Options for internal evaluation: Quizzes, Presentations, Surveys, Internship, Tutorials, Role Play,